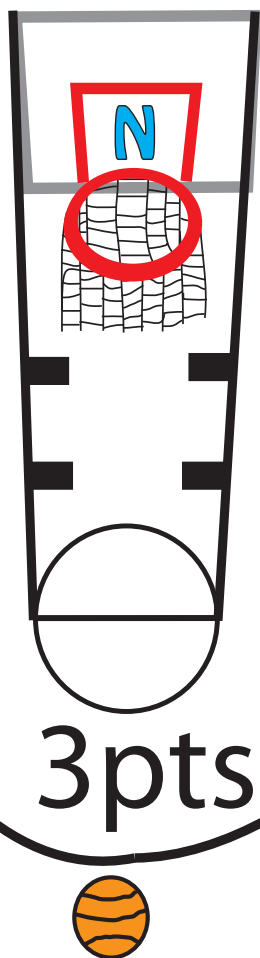
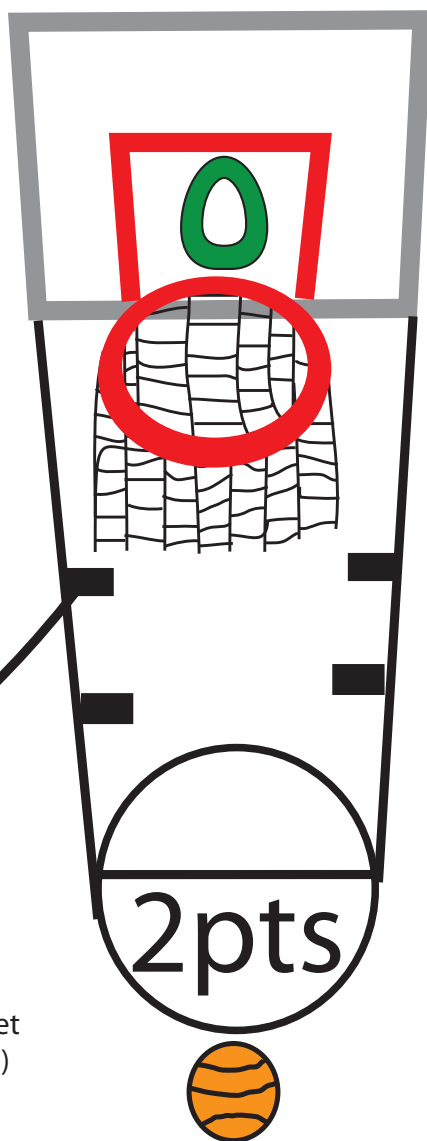


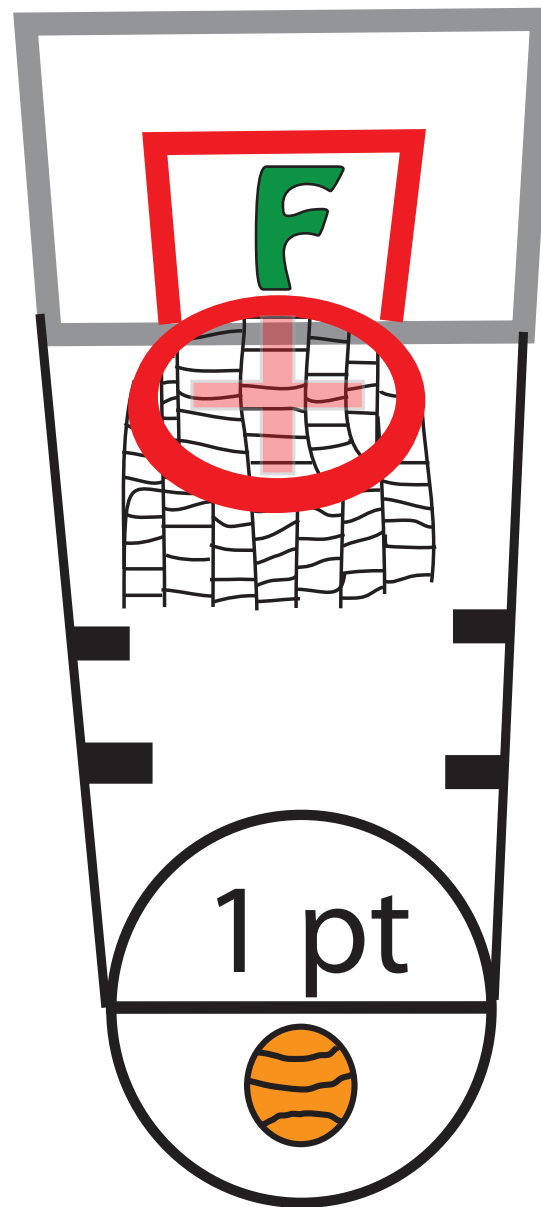
Periodic Tables of Valence (e-) Electron Basket Ball Basket 'Schroeder Vacancies'



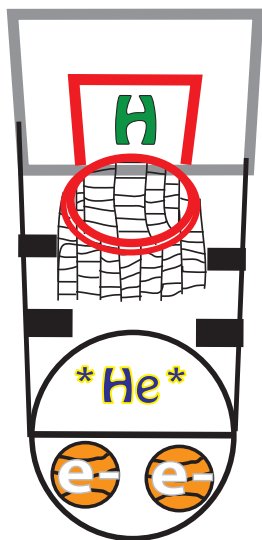
Nitrogen would be a Three Pointer
(N needs three e-s to balance for octet
= eight Valence like Neon (and Stable))



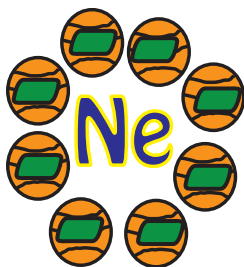
Oxygen is a simple Two-Pointer
so get Two e-s from the
'Top of the Key'



F for Free Throw
is only One Point
so is ONE e-



Make TWO of the H's
... have H₂ or He ...



ex. NH₃, H₂O, HF

You need the valence e-s to resemble the octet of valence (outermost shell)
which has eight e-s (like the closest Noble Gas) so one, two, three pointers...